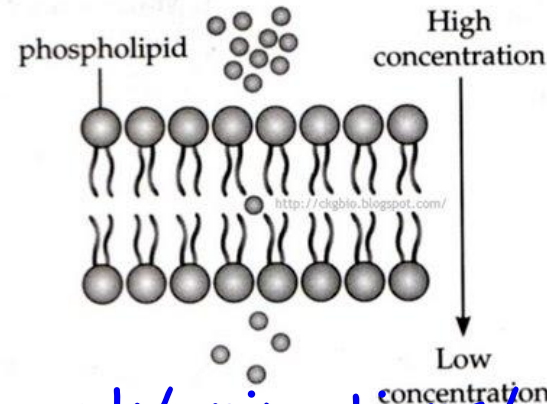


Passive Transport Mechanisms

Aim: To understand how water potential affects cell tonicity

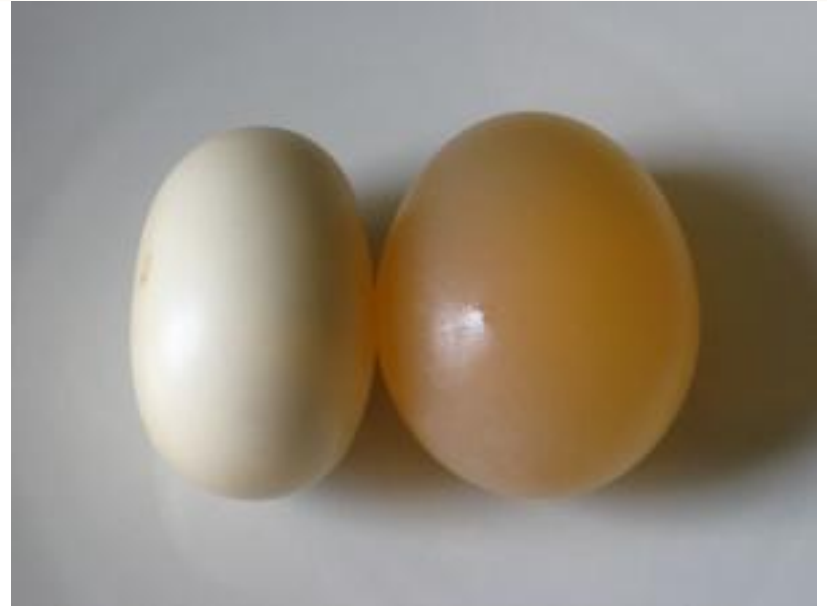


http://www.kscience.co.uk/animations/water_potential.swf

Finishing the Egg-Speriment

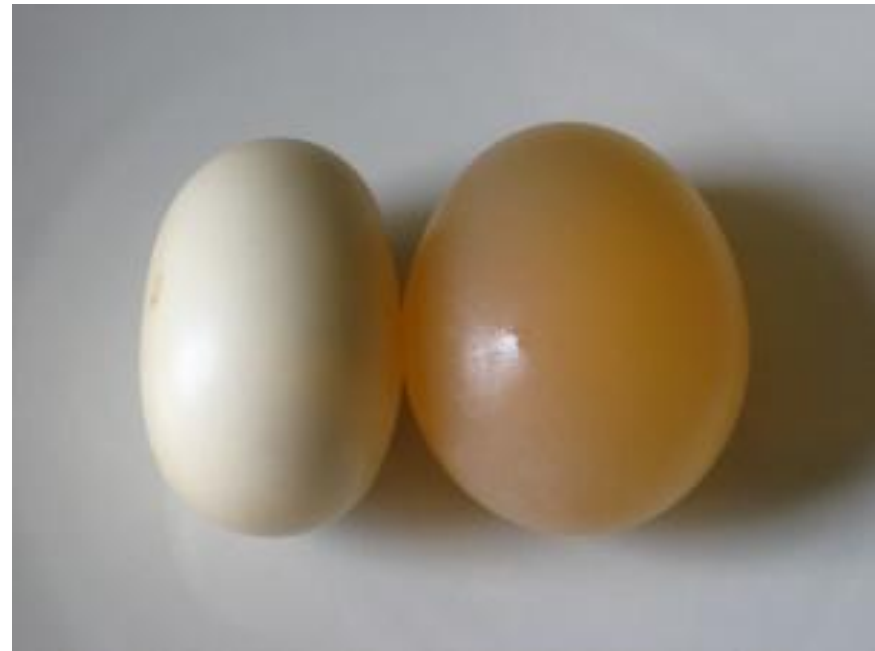
- CAREFULLY wash eggs.
- Weigh eggs & record data (*quantitative*) on a sheet that can be handed to Ms for data input
- Line up eggs with everyone else's. Can you see a differences?
- Take *qualitative* notes/observations

Egg photo shoot

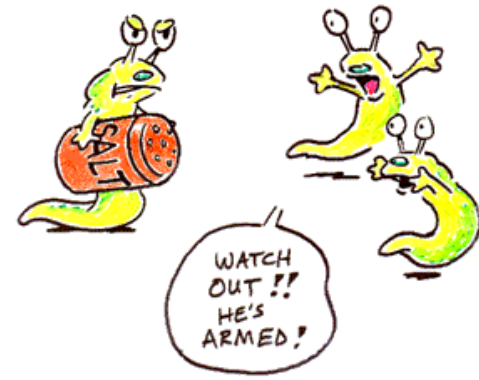


Coursework

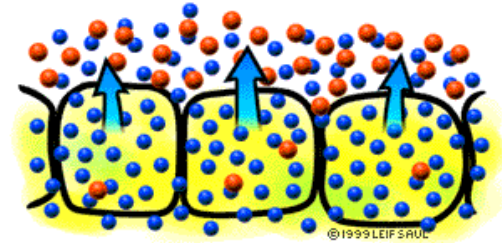
- WJEC CW - Completed in exam style conditions
- 3 sections of coursework to be completed *by hand*
- Table, Graph (no extrapolation!), qualitative data, analysis of results and further work (research!!!)
- Due next lesson
- 2 hours minimum



Osmosis

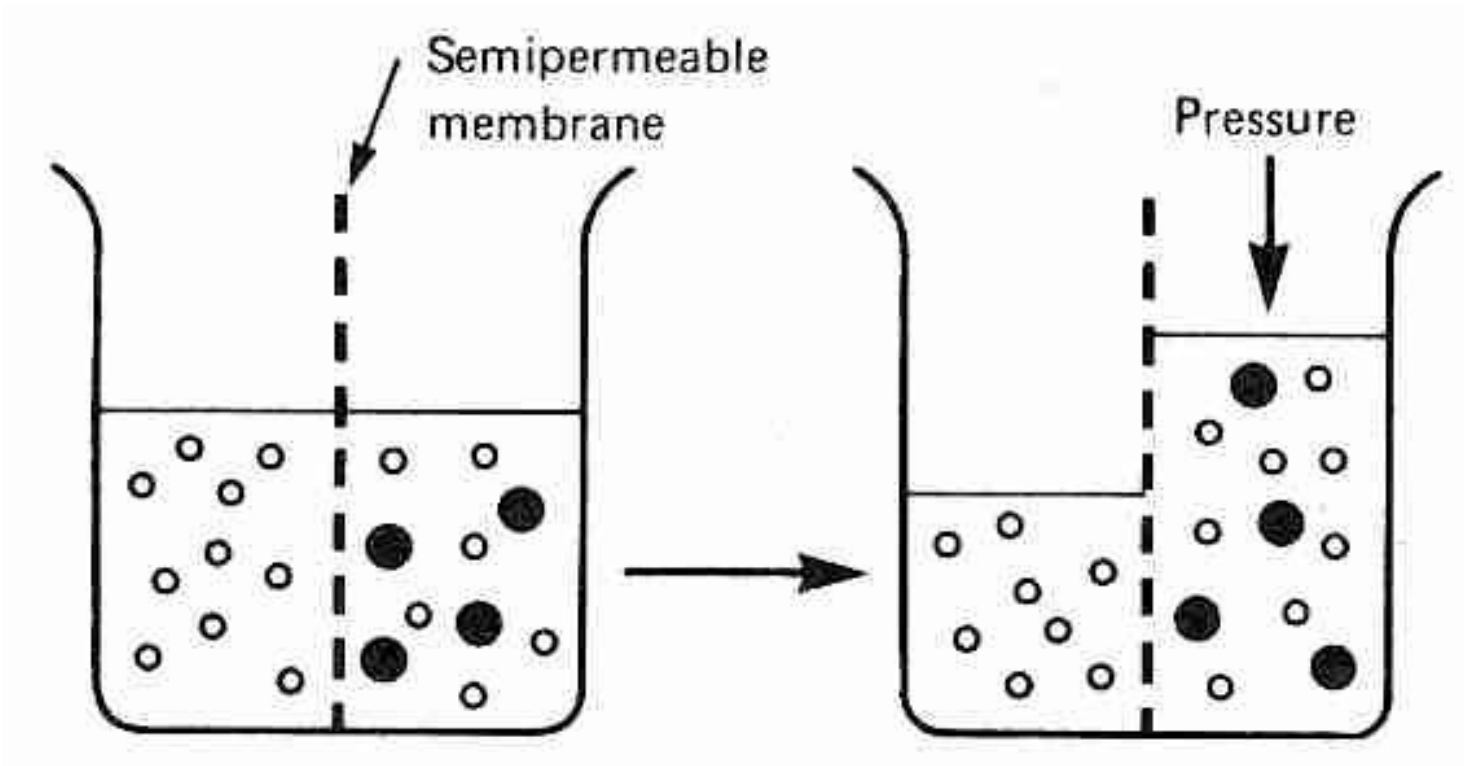


Learning Objectives



1. Describe passive processes of movement across cell membranes (C)
2. Explain water potential using osmosis and Ψ_s / Ψ_p
3. Explain the effects of hyper - hypo and iso-tonic solutions on The Naked Egg.

Explain what has happened to the water in terms of Ψ_s and Ψ_p



Right side has lower/more negative water potential (Ψ_s). As water moves to an area of lower water potential water move to the right until $\Psi_p + \Psi_s$ on the right is equal to that of the left.

Effect of solution concentration on Cells (RBCs)

- **Hypotonic Solution**

Hypo = less solute than cell.

Water drawn INTO the cell

- **Isotonic Solution**

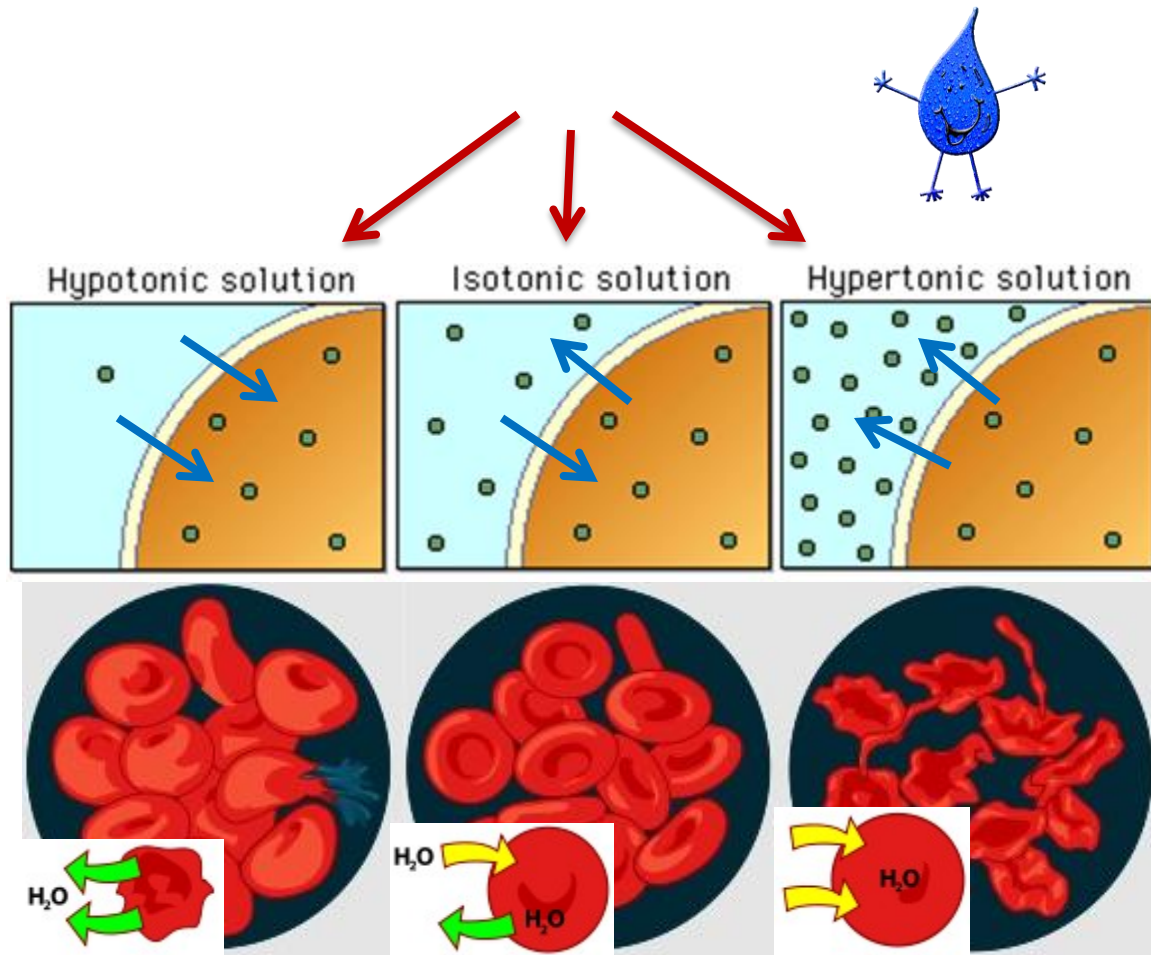
Iso = same amount of solute

No change - Dynamic Equilibrium

- **Hypertonic Solution**

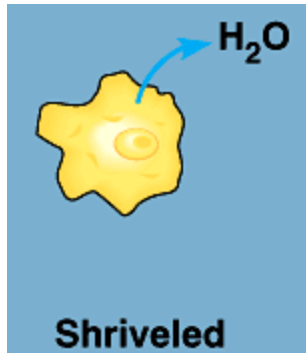
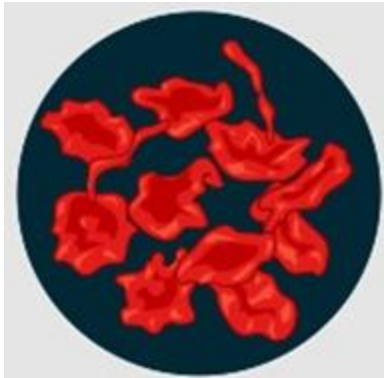
Hyper = more solute than cell.

Water drawn OUT of the cell

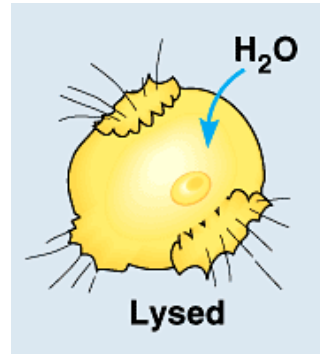


Decreasing water potential of solution cells are suspended in. $WP = 0$ $WP = -$

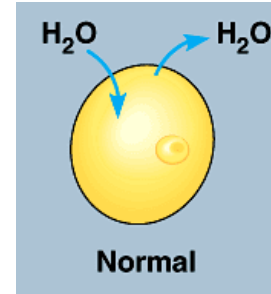
Is the solution hypo, iso, or hyper-tonic to the cell?



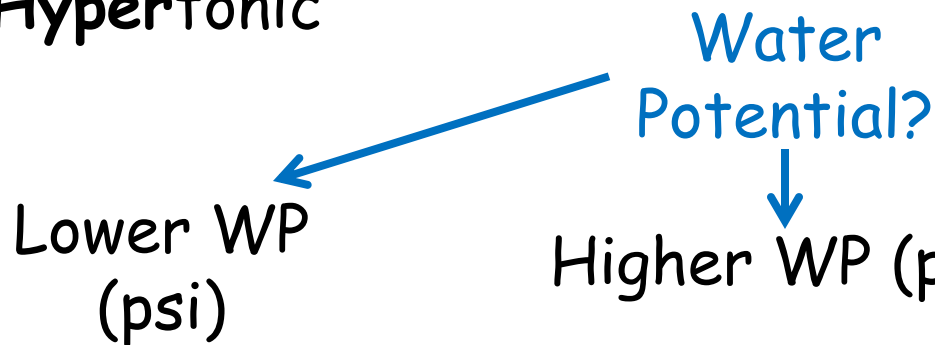
Hypertonic



Hypotonic



Isotonic



Name:

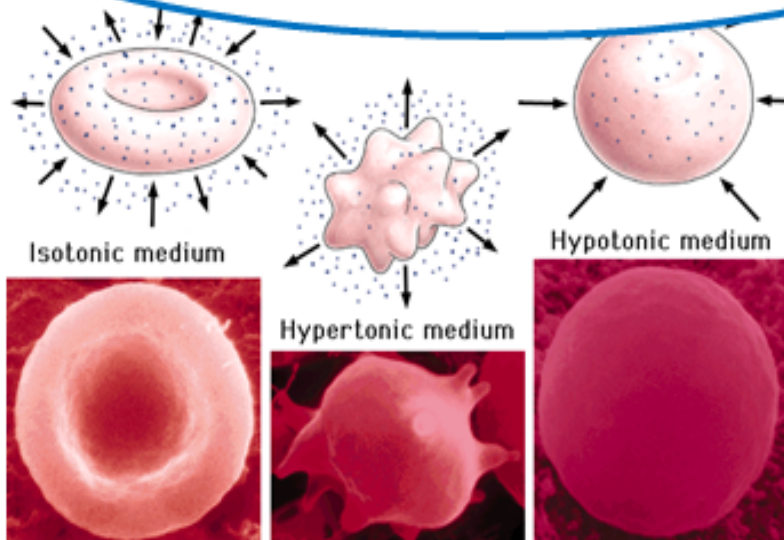
Set:

Osmosis & Water Potential

Define osmosis in terms of water potential:

Osmosis is the way water moves from an area of high water potential or high Ψ to an area of low water potential or low Ψ , through a selectively-permeable membrane.

What is Ψ ?



Using water, explain why each red blood cell looks like it does:

Isotonic— Isotonic cell looks like this because it hasn't gained or lost water. Equal water pot

Hypertonic— Hypertonic cell necrotises as it's lost water due to solution of higher water pot.

Hypotonic - Hypotonic cell is swelled as it has gained water in solution of lower water potential

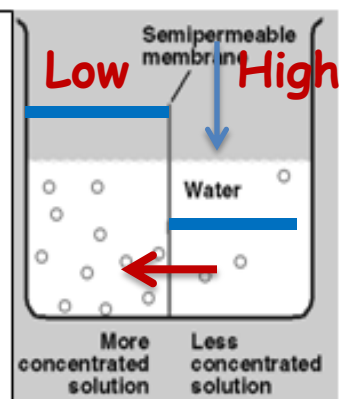
What is Ψ in an animal cell?

Solute potential only: Ψ_s

1. Draw what will happen to the water in this beaker.
2. Draw arrows to show the direction the water moves.
3. Label which sides have high/low water potential.

4. Add an arrow to show where Ψ_p will affect the solution.

5. Describe how the water moves:



Q1: Where have you heard the word isotonic before?

Sports drinks?

Q2: Why do you think this is beneficial for sports drinks?

Fast uptake of glucose without dehydrating

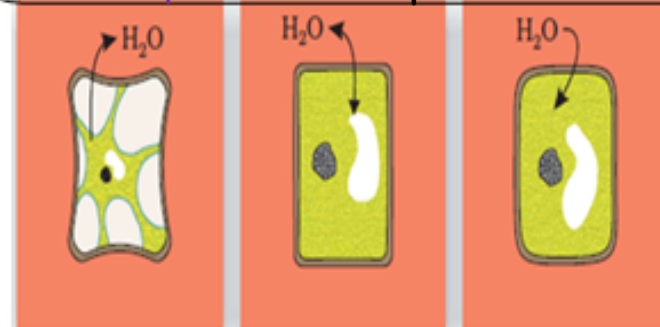
Q3: What is the difference between putting an animal cell and a plant cell in to hypo and hypertonic solutions?

Hypotonic: Plant cell loses water, membrane pulls away from cell wall (plasmolysed). RBC necrotises

Hypertonic: Plant cell become turgid due to Ψ_p of cell wall, animal cell bursts - lyses.

Q4: What is Ψ equal to in a plant cell? Solute potential

+ Pressure potential: $\Psi_s + \Psi_p$



Plasmolysed

Turgid



Water Potential

Water moves from an area of high wp to an area of low wp.

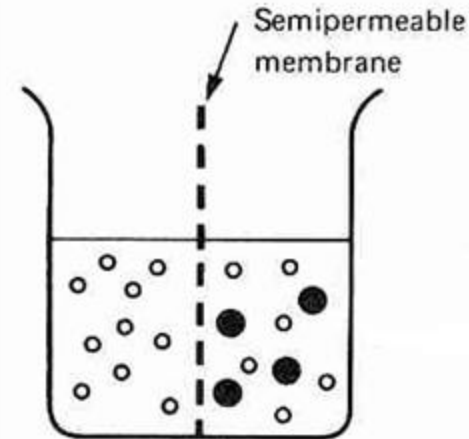
1.

What must we do to equalise this U bend?



2.

Which side has lower Ψ ?



3.

Using Ψ , explain how the **water** will move?

